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TITLE UFe4P12 and CeFe4P12: Non-Metallic Isotypes of Superconducting LaFe<sub>4</sub>P<sub>12</sub>

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## UFe,P12 AND CEFE,P12: MON-METALLIC ISOTYPES OF SUPERCONDUCTING LaFe,P12

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The new compound UFe<sub>4</sub>P<sub>12</sub>, which was found to be isostructural to superconducting LaFe<sub>4</sub>P<sub>12</sub> and with a lattice constant of 7.7729 Å, is a semiconductor and shows ferromagnetic order below 3.15 K. CeFe<sub>4</sub>P<sub>12</sub> is also a semiconductor and its magnetic susceptibility is unusually small in comparison to LaFe<sub>4</sub>P<sub>12</sub>. The semiconducting behaviors of both UFe<sub>4</sub>P<sub>12</sub> and CeFe<sub>4</sub>P<sub>12</sub> seem anomalous and may arise from strong f-electron hybridization.

72.80.Jc, 75.50.Dd, ?5.30.Cr

As part of our investigation of various types of f-electron materials, we have included the study of a system of ternary iron phosphide compounds with the general formula  $MFe_{L}P_{1,2}$ , where M can be the elements La through Eu (1), Th (2) or U. The compound LaFe P12 is a superconductor with a transition temperature  $T_c = 4.1 \text{ K}$  and exhibits normal metallic conductivity with a residual resistance ratio of about 90(3,4,5) Metallic conductivity is also observed for PrFe<sub>4</sub>P<sub>12</sub> (5) and NdFe<sub>4</sub>P<sub>12</sub> (6) which undergo magnetic transitions at low temperatures. (3) Mossbauer experiments (7) and magnetic susceptibility measurements (4) on LaFe<sub>4</sub>P<sub>12</sub> are consistent with completely demagnetized iron in these compounds, suggesting that the properties of members of the series are determined by the M constituent. In this paper we report the preparation of UFe,P,2 for the first time, and report measurements of the electrical resistivity and magnetic susceptibility of both UFe4P12 and CeFe4P12. Our results imply that delocalized f-electrons are present in these two compounds and are strongly hybridized with electrons of the surrounding iron-phosphorous sublattice. This leads to a significant departure of their electronic properties from those expected based on the other MFe4P12 compounds.

Measurements of UFe $_4P_{12}$  and CeFe $_4P_{12}$  were performed on single crystal specimens that were grown in a molten tin solvent. E1-ectrical resistivities were measured using a standard 4-wire technique. The magnetic susceptibility of CeFe $_4P_{12}$  was measured with a Faraday magnetometer in 8531 gauss applied magnetic fields and of UFe $_4P_{12}$  with a SQUID magnetometer in 10 kgauss. The body-centered

cubic crystal structure (1) of UFe<sub>4</sub>P<sub>12</sub> and CeFe<sub>4</sub>P<sub>12</sub> was verified using x-ray diffraction analysis.

The electrical resistivities of  $UFe_4P_{12}$  and  $CeFe_4P_{12}$  are shown in Fig. 1. The curves are normalized to the room temperature values which we estimate to be greater than 1000 μΩ-cm, although precise values could not be determined due to the irregular shape of the crystals. For both, the resistivity rise with decreasing temperature by over six orders of magnitude demonstrates clearly that they are semiconductors. To our knowledge this is the first example of an isostructural series of metallic rare earth compounds with the Ce member being non-metallic. Both UFe4P12 and CeFe4P12 can be fit only over the limited temperature range 85K<T<140K to an activated conduction form,  $R=R\exp\left(\Delta E/k_{R}T\right)$ , where  $\Delta E$  is the activation energy and  $k_{\rm B}$  is Boltzmann's constant. The values  $\Delta E/k_R = 360K$  and 1500K, respectively, are obtained. Below 85K both curves show a break, or knee, and can no longer be described by simple activated... conduction. In the lower temperature, higher resistivity range the measuring current was varied as a check that there was no resistive heating of the samples.

From crystal chemical arguments  $^{(1,9)}$  it was expected that  $\operatorname{CeFe}_{\Lambda}\operatorname{P}_{12}$  could be a semiconductor. This was based on the analogy of these phosphides with the semiconducting compound  $\operatorname{CoP}_3$ , which is isostructural to the iron phosphorus sublattice in the  $\operatorname{MFe}_4\operatorname{P}_{12}$  structure.  $^{(10)}$  if  $\operatorname{Ce}$  were tetravalent, it would contribute the necessary four conduction electrons to the Iron-phosphorus sublattice, thus achieving the same electron concentration as  $\operatorname{CoP}_3$ . In

 $LaFe_4P_{12}$ , which is a good metal, the rare earth ion is trivalent. If Ce were to be nearly trivalent in the MFe4P12 phase, one would expect metallic (or perhaps Kondo-like) conductivity and a strongly temperature dependent susceptibility. The fact that CeFe P12 is a semiconductor and shows very small, nearly temperature independent susceptibility (see below) would indicate a nearly tetravalent state for Ce or perhaps strong 4f electron hybridization. However, this argument is weakened somewhat when a comparison is made with ThFe4P12 where Th is almost certainly tetravalent. Our preliminary resistivity measurements show poor. metallic conductivity for ThFe<sub>4</sub>P<sub>12</sub>. In UFe<sub>4</sub>P<sub>12</sub> semiconducting behavior is accompanied by localized moment type magnetism. Thus this system may possess two localized 5f electrons, making the U ions in this system analogous to Pr. It seems likely that the itinerant f-electrons present in  $UFe_4P_{12}$  and  $CeFe_4P_{12}$  strongly hybridize with the conduction electrons and open a gap in the electronic density of states at the fermi energy. This may be analogous to the hybridization gap seen in the intermediate valent systems SmB and SmS; except that these materials have considerably smaller activation energies than  $CeFe_{L}P_{12}$  and  $UFe_{L}P_{12}$ . Finally, we see that the lattice constants for  $UFe_4P_{12}$  and  $CeFe_4P_{12}$  (7.7729 Å and 7.7920 Å, respectively) are the smallest known for these types of compounds, and this lends credence to the argument for hybridization.

The magnetic susceptibility of two samples of  $CeFe_4P_{12}$  are shown in Fig. 2 along with that of  $LaFe_4P_{12}$  for comparison. The upturns

at low temperature for  $\text{CeFe}_4P_{12}$  can be fit quite well to a Curie law which varies by an order of magnitude between the two batches of  $\text{CeFe}_4P_{12}$  crystals. The temperature independent contribution to the susceptibility of both batches is nearly the same. We therefore attribute the upturns to impurities, which are most likely second phases on the surface of the crystals, although some part of the upturn of the second batch may be intrinsic. The susceptibility of the second batch can be characterized by a temperature independent term  $\chi_0 = 2.6 \times 10^{-4} \text{ cm}^3/\text{mole}$  and a Curie term equivalent to 0.3%  $\text{Ce}^{3+}$  impurity ions per formula unit. Thus the susceptibility of  $\text{CeFe}_4P_{12}$  is roughly a factor of two smaller than that of  $\text{LaFe}_4P_{12}$ , and the occurrence of a Ce compound having a much smaller susceptibility than the isostructural La compound is a rather unusual result.

The magnetic behavior of UFe<sub>4</sub>P<sub>12</sub> as seen from Fig. 3 is very different from that of CeFe<sub>4</sub>P<sub>12</sub>. A ferromagnetic-like transition at 3.15 K was inferred from ac susceptibility measurements of UFe<sub>4</sub>P<sub>12</sub>. In addition, the magnetization in fields up to 3 T showed strong field dependence and attained 1.2  $\mu_{\rm B}$ /U-atom at T=1.9 K. Magnetic hysteresis was detected for temperatures below the Curie temperature. The small positive Curie-Weiss temperature in Fig. 3 is consistent with low temperature ferromagnetic order. We note that the ferromagnetic order cannot be mediated by conduction electrons, as UFe<sub>4</sub>P<sub>12</sub> is a semiconductor. The inverse susceptibility vs temperature data of Fig. 3 deviate somewhat from a Curie-Weiss law at high temperature and give an effective moment of 2.25  $\mu_{\rm B}$ /U-atom at low temperatures. This moment

could be accounted for by a 5f<sup>2</sup> configuration reduced by crystalline electric field effects or hybridization.

In summary, the properties of the compound UFe<sub>4</sub>P<sub>12</sub> are reported for the first time. Our results presented above on the resistivity and magnetic susceptibility of CeFe<sub>4</sub>P<sub>12</sub> and UFe<sub>4</sub>P<sub>12</sub> are highly unusual in comparison to the other isostructural members of this class of compounds. Strong hybridization of the f-electrons of the Ce and U atoms with the conduction electrons is likely to be responsible for this unusual behavior.

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## FIGURE CAPTIONS:

- Fig. 1: The electrical resistances, R, of UFe<sub>4</sub>P<sub>12</sub> and CeFe<sub>4</sub>P<sub>12</sub> normalized to the room temperature resistance  $R_{RT}$ .
- Fig. 2: The magnetic susceptibility of CeFe<sub>4</sub>P<sub>12</sub> for two batches of crystals. The susceptibility of LaFe<sub>4</sub>P<sub>12</sub> is shown for comparison.
- Fig. 3: The inverse magnetic susceptibility of UFe<sub>4</sub>P<sub>12</sub>. The solid line represents an effective moment of 2.25  $\mu_B/U$ -atom.

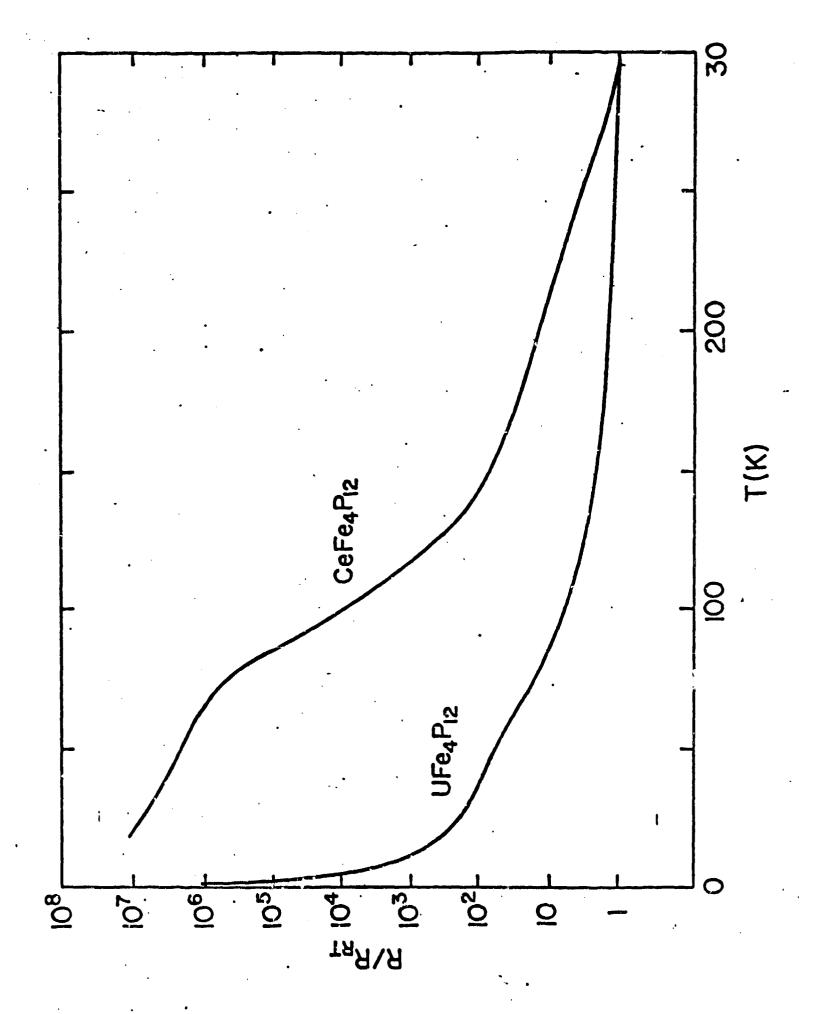
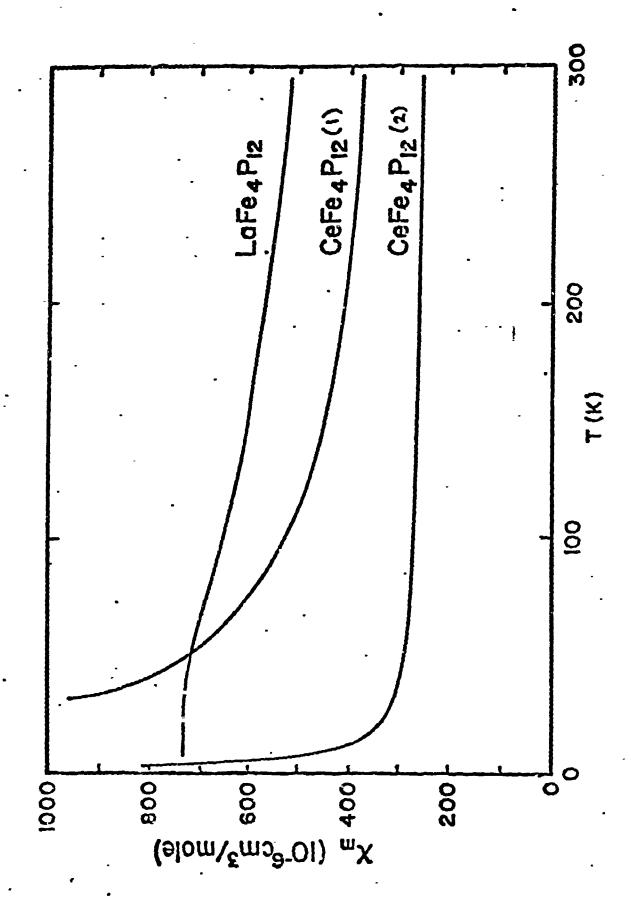


Figure 1



Flyure 2

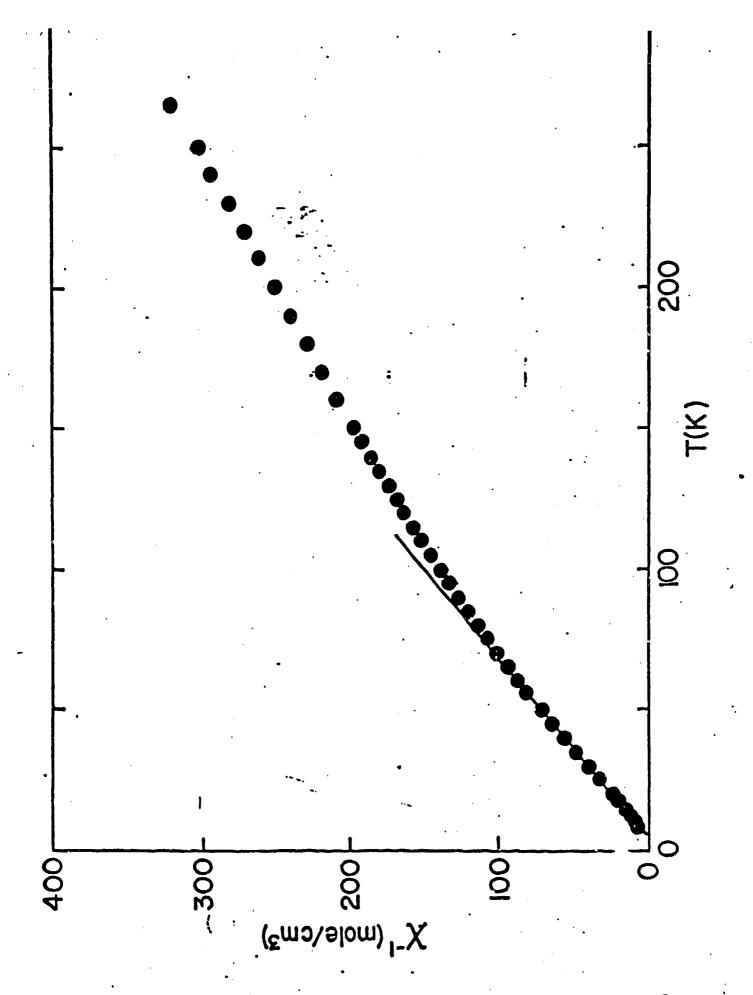


Figure 3